

This article was downloaded by:

On: 30 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Spectroscopy Letters

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713597299>

Level Crossings in Indo Bending Potentials for CO₂

F. A. Van-catledge^a

^a Department of Chemistry, University of Minnesota, Minneapolis, Minnesota

To cite this Article Van-catledge, F. A.(1974) 'Level Crossings in Indo Bending Potentials for CO₂', Spectroscopy Letters, 7: 8, 405 — 408

To link to this Article: DOI: 10.1080/00387017408067265

URL: <http://dx.doi.org/10.1080/00387017408067265>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

LEVEL CROSSINGS IN INDO BENDING POTENTIALS FOR CO₂KEY WORDS: INDO, CONFIGURATION INTERACTION

F. A. Van-Catledge

Department of Chemistry
University of Minnesota
Minneapolis, Minnesota 55455

Recently, Combs and Holloman have reported anomalous energy minima in approximate SCF calculations on bending modes in several small molecules.^{1,2} The methods employed (CNDO/2 and INDO)³ are currently in wide use and it would be helpful to have a clear view of the origin(s) of these anomalies. [As has been demonstrated,² reparameterization to remove these features creates more serious problems than that for which the remedy was derived.] We were particularly concerned as recent studies have shown that INDO wavefunctions, when properly handled, provide a qualitatively correct description of the electronic charge distribution in selected diatomic molecules.⁴ Further, these methods are extensively used for studies of molecular deformations.⁵

On examining the published curves for CO₂ and FNO₂, we noted what appeared to be discontinuities in the first derivative for the energy vs bond angle plots in the "barrier" region. This suggested that the single

determinantal wavefunction is, for some reason, inadequate at this level of approximation. We report herein results which support this point of view. We have developed in our laboratory the program INDOCI2 which, subsequent to the SCF calculation, generates a set of configuration interaction wavefunctions built from all doubly excited configurations involving pairwise excitation, e.g., for the ground state

$$\Psi_{CI}^0 = \Phi_{SCF} + \sum_{i,k} C_{ik} \Phi_{ii}^{kk} \quad (1)$$

Also available are the SCF configuration energies calculated as

$$\langle \Phi_{ii}^{kk} | \hat{H} | \Phi_{ii}^{kk} \rangle - E^0 = 2(\epsilon_k - \epsilon_i) - 4J_{ik} + 2K_{ik} + J_{kk} + J_{ii} \quad (2)$$

Using this method we have re-examined the bending coordinate for CO_2 ($r_{\text{CO}} = 1.1612 \text{ \AA}$) over the range $60^\circ \leq \theta \leq 180^\circ$.

It is important to monitor orbital occupation as a function of bending. In the region 95° - 180° , INDO calculations give the ordering under the point group $D_{\infty h}$

$$3a_{1g}^2(3a_1^2), \quad 2a_{2u}^2(2b_1^2), \quad 1e_{1u}^4(1b_2^2, 4a_1^2), \quad 4a_{1g}^2(5a_1^2), \\ 3a_{2u}^2(3b_1^2), \quad 1e_{1g}^4(1a_2^2, 3b_1^2); \quad 2e_{1u}(6a_1, 2b_2) \quad .$$

The parenthesized specifications correspond to the symmetry designations appropriate for $\theta \neq 180^\circ$ under the point group C_{2v} . In our calculations we found that the ground state (SCF) electronic configuration was $(\dots 1a_2^2 3b_1^2)$ in the region 95° - 180° , but changed to $(\dots 1a_2^2 6a_1^2)$ at smaller angles. Clearly, then, the aforementioned discontinuity in slope arises via level crossing (see Fig. 1). Inspection of the behavior of the relevant configuration energies (before CI) reveals curve discontinuities in the region of the maximum. This arises from the use of ground state SCF orbitals to calculate their energies (eqn. 2). The relief of these dis-

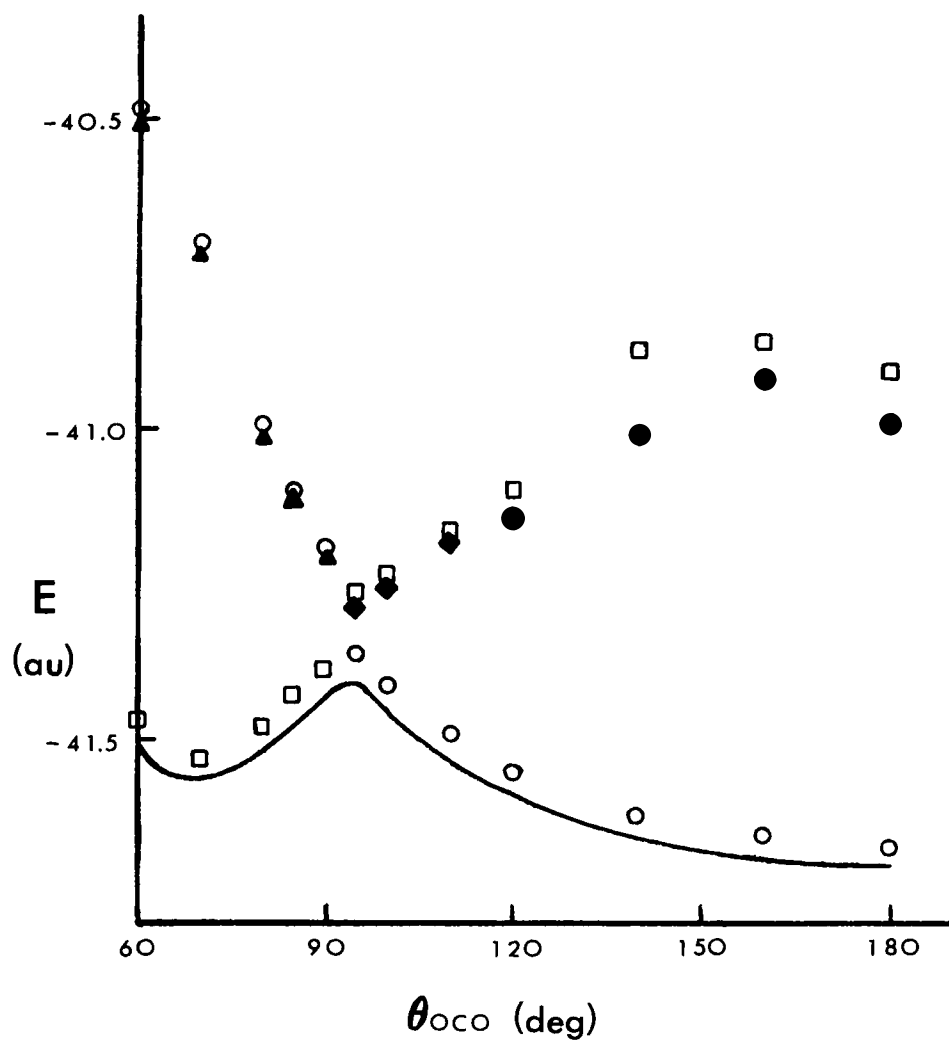


FIG.

THE TOTAL ENERGY OF SELECTED ELECTRONIC STATES FOR CO₂ VERSUS THE OCO ANGLE.
 (□) ($\cdots 1a_2^2 6a_1^2$) BEFORE CI; (◆) ($\cdots 1a_2^2 6a_1^2$) AFTER CI; (○) ($\cdots 1a_2^2 3b_1^2$) BEFORE CI; (▲) ($\cdots 1a_2^2 3b_1^2$) AFTER CI; (—) ψ^0 (AFTER CI).

continuities is not apparent after CI. This is due in part to the plethora of avoided crossings in this region. Several configurations not shown, in particular ($\dots 4b_1^2 6a_1^2$), make strong contributions in this regard. The important feature to note, however, is that the slope discontinuity for the ground state wavefunction has been removed by this procedure.

It is possible that more extensive CI might eliminate the (presumably) spurious maximum altogether, but this begs the main questions regarding INDO calculations. The recent non-empirical SCF-MO studies of CO_2 by Davidson, et. al., show no level crossing in the region 90° - 180° .⁶ This is completely analogous to the results for FNO_2 .⁷ Clearly, then, spurious level crossing arises as a consequence of the approximations inherent in CNDO and INDO. The recent INDO studies of allene deformation⁵ show a local minimum which, in view of the results reported herein, is suspect. It seems clear that these NDO methods, while suitable for small molecular deformations, can lead to erroneous conclusions unless employed judiciously.

REFERENCES

1. L. L. Combs and M. Holloman, *Spectroscopy Letters*, 5, 319 (1972).
2. L. L. Combs and M. Holloman, *Spectroscopy Letters*, 6, 257 (1973).
3. a) J. A. Pople, D. L. Beveridge, and P. A. Dobush, *J. Chem. Phys.*, 47, 2026 (1967); b) J. A. Pople and G. A. Segal, *ibid.*, 44, 3289 (1966).
4. F. A. Van-Catledge, *J. Phys. Chem.*
5. See, for example, P. W. Dillon and G. R. Underwood, *J. Amer. Chem. Soc.*, 96, 779 (1974).
6. L. Z. Stenkamp and E. R. Davidson, *Theoret. Chim. Acta*, 30, 283 (1973).
7. L. Burnelle, A. M. May, and R. A. Gangi, *J. Chem. Phys.*, 49, 561 (1968).

Received June 28, 1974

Accepted July 2, 1974